# Nanocrystals in the Kitchen 

Thomas Fernique

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## Reformulation:

What is the maximal density of a packing of disks of given sizes?
Our running example:
Disks of diameter 1 and $r:=\sqrt{2}-1$ :


## First try



This ratio allows a small disk to exactly fit between four large ones.

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This packing has density $\delta^{*}:=\frac{\pi+\pi r^{2}}{4} \geq 92 \%$. Optimal?

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The Hexagonal Compact Packing (HCP) has density $\frac{\pi}{2 \sqrt{3}} \leq 91 \%$.

## Proof intuition

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The densest triangle which connects the centers of equal disks connects three pairwise adjacent disks.

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Proof idea: "spread" the density to lower it everywhere below $\delta^{*}$.

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4. show, for every triangle $T$ of $\mathcal{T}$ :

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\begin{equation*}
\sum_{v \in T} U_{v}(T) \leq E(T) \tag{2}
\end{equation*}
$$

where $E(T):=\delta^{*} \operatorname{vol}(T)-\operatorname{cov}(T)$ is the emptiness of $T$.

## Step 1: Delaunay triangulation



Points closer to a disk than to any other one $\rightsquigarrow$ Voronoï partition.

## Step 1: Delaunay triangulation



Dual of the Voronoï partition $\rightsquigarrow$ Delaunay triangulation.

## Step 1: Delaunay triangulation



Partition vertex: center of a disk interior-disjoint from the packing.

## Step 1: Delaunay triangulation



Claim: saturation $\Rightarrow$ edge lengths $\leq 2+2 r$ and angles $\geq 33^{\circ}$.

## Step 2: Vertex potential



If $T$ connect the centers $u, v$ and $w$ of disks of size $r_{u}, r_{v}$ and $r_{w}$ :

$$
U_{v}(T):=U_{r_{u} r_{v} r_{w}}^{*}+m\left|\hat{v}-\hat{v}^{*}\right|
$$

where:

- $\hat{v}$ is the angle in $v$ of $T$ and $\hat{v}^{*}$ in its tight version $T^{*}$;
- $U_{r_{u} r_{v} r_{w}}^{*}=U_{r_{w} r_{v} r_{u}}^{*} \in \mathbb{R}$ is the base vertex potential (to be fixed);
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We want:
$\sum_{T \ni v} U_{v}(T) \geq 0$.


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## Step 4: Inequality (2)

We also want:

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\sum_{v \in T} U_{v}(T) \leq E(T)=\delta^{*} \operatorname{vol}(T)-\operatorname{cov}(T)
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Use the Mean Value Theorem (and, again, Interval Arithmetic).

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- The case of spheres of size 1 and $\sqrt{2}-1$ should be easier!
- Motivation: material sciences (nanocrystals).



